

Introduction Chemistry Lab Manual Corwin

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this is the latest version of charles h corwin s best selling lab manual which has been used by over 250 000 students this fourth edition retains the highly effective format of a prelaboratory assignment a stepwise procedure and a postlaboratory assignment these experiments have been used successfully with students enrolled in introductory chemistry preparatory chemistry and allied health chemistry

paying particular attention to the environmental issue the fifth edition of this popular chemistry lab manual retains an effective format of a prelaboratory assignment a stepwise procedure and a postlaboratory assignment introduction to chemistry instrumental measurements density of liquids and solids freezing points and melting points physical properties and chemical properties atomic fingerprints families of elements identifying cations in solution identifying anions in solution analysis of a penny determinations of avogadro s number empirical formulas of compounds analysis of alum decomposing baking soda precipitating calcium phosphate generating hydrogen gas generating oxygen gas molecular models and chemical bonds analysis of saltwater analysis of vinegar electrical conductivity of aqueous solutions activity series of metals organic models and functional groups separation of food colors and amino acids a useful reference for professionals in the allied health chemistry fields

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this newest version of laboratory activities has evolved from charles h corwin s experiments which have been used by nearly 200 000 students in addition to the fresh new art program that enhances student orientation to each experiment this version retains the highly successful format of prelaboratory preparation stepwise guided procedures and postlaboratory assignments the laboratory manual is especially well suited for students in introductory chemistry preparatory chemistry and allied health chemistry in this newest version the changes and improvements include particular attention to the environmental issue this version does not contain any procedures involving lead mercury chromium chloroform or carbon tetrachloride experiments that utilize 13 x 100 mm test tubes rather than 16 x 150 mm test tubes so as to further reduce chemical waste no special equipment is required and the labs are not microscale an increased effort to ensure the safety of students in the laboratory operations that involve even minimal potential danger have been avoided students are alerted to procedures that should be performed carefully and the prelaboratory assignments have questions regarding safety example exercises that illustrate the calculations associated with quantitative experiments earlier placement of chemical reactions to motivate students while experiencing highly visual observations and color changes

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