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advanced thermodynamics engineering second edition is designed for readers who need to understand and apply the engineering physics of thermodynamic concepts it employs a self teaching format that reinforces presentation of critical concepts mathematical relationships and equations with concrete physical examples and explanations of applications to help readers apply principles to their own real world problems less mathematical theoretical derivations more focus on practical application because both students and professionals must grasp theory almost immediately in this ever changing electronic era this book now completely in decimal outline format uses a phenomenological approach to problems making advanced concepts easier to understand after a decade teaching advanced thermodynamics the authors infuse their own style and tailor content based on their observations as professional engineers as well as feedback from their students condensing more esoteric material to focus on practical uses for this continuously evolving area of science this book is filled with revised problems and extensive tables on thermodynamic properties and other useful information the authors include an abundance of examples figures and illustrations to clarify presented ideas and additional material and software tools are available for download the result is a powerful practical instructional tool that gives readers a strong conceptual foundation on which to build a solid functional understanding of thermodynamics

thermal and statistical physics has established the principles and procedures needed to understand and explain the properties of systems consisting of macroscopically large numbers of particles by developing microscopic statistical physics and macroscopic classical thermodynamic descriptions in tandem statistical and thermal physics an introduction provides insight into basic concepts and relationships at an advanced undergraduate level this second edition is updated throughout providing a highly detailed profoundly thorough and comprehensive introduction to the subject and features exercises within the text as well as end of chapter problems part i of this book consists of nine chapters the first three of which deal with the basics of equilibrium thermodynamics including the fundamental relation the following three chapters introduce microstates and lead to the boltzmann definition of the entropy using the microcanonical ensemble approach in developing the subject the ideal gas and the ideal spin system are introduced as models for discussion the laws of thermodynamics are compactly stated the final three chapters in part i introduce the thermodynamic potentials and the maxwell relations applications of thermodynamics to gases condensed matter and phase transitions and critical phenomena are dealt with in detail initial chapters in part ii present the elements of probability theory and establish the thermodynamic equivalence of the three statistical ensembles that are used in determining probabilities the canonical and the grand canonical distributions are obtained and discussed chapters 12 15 are concerned with quantum distributions by making use of the grand canonical distribution the fermi dirac and bose einstein quantum distribution functions are derived and then used to explain the properties of ideal fermi and bose gases the planck distribution is introduced and applied to photons in radiation and to phonons on solids the last five chapters cover a variety of topics the ideal gas revisited nonideal systems the density matrix reactions and irreversible thermodynamics a flowchart is provided to assist instructors on planning a course key features fully updated throughout with new content on exciting topics including black hole thermodynamics heisenberg

~~antiferromagnetic chains entropy and information theory renewable and nonrenewable energy sources and the mean field theory of antiferromagnetic systems additional problem exercises with solutions provide further learning opportunities suitable for advanced undergraduate students in physics or applied physics michael j r hoch spent many years as a visiting scientist at the national high magnetic field laboratory at florida state university usa prior to this he was a professor of physics and the director of the condensed matter physics research unit at the university of the witwatersrand johannesburg where he is currently professor emeritus in the school of physics~~

this introductory textbook for standard undergraduate courses in thermodynamics has been completely rewritten to explore a greater number of topics more clearly and concisely starting with an overview of important quantum behaviours the book teaches students how to calculate probabilities in order to provide a firm foundation for later chapters it introduces the ideas of classical thermodynamics and explores them both in general and as they are applied to specific processes and interactions the remainder of the book deals with statistical mechanics each topic ends with a boxed summary of ideas and results and every chapter contains numerous homework problems covering a broad range of difficulties answers are given to odd numbered problems and solutions to even numbered problems are available to instructors at cambridge.org/9781107694927

this is a textbook on thermodynamics of materials for junior senior undergraduate students and first year graduate students as well as a reference book for researchers who would like to refresh their understanding of thermodynamics the textbook employs a plain language to explain the thermodynamic concepts and quantities it embraces the mathematical beauty and rigor of gibbs thermodynamics through the fundamental equation of thermodynamics from which all thermodynamic properties of a material can be derived however a reader with basic first year undergraduate calculus skills will be able to get through the book without difficulty one unique feature of this textbook is the descriptions of the step by step

~~procedures for computing all the thermodynamic properties from the fundamental~~
equation of thermodynamics and all the thermodynamic energies from a set of
common experimentally measurable thermodynamic properties supplemented with
ample numerical examples another unique feature of this textbook is its emphasis
on the concept of chemical potential and its applications to phase equilibria in
single component systems and binary solutions chemical reaction equilibria and
lattice and electronic defects in crystals the concept of chemical potential is
introduced at the very beginning of the book together with temperature and
pressure it avoids or minimizes the use of terms such as molar gibbs free energy
partial molar gibbs free energy or gibbs potential because molar gibbs free energy
or partial molar gibbs free energy is precisely the chemical potential of a material or
a component it is the chemical potential that determines the stability of chemical
species compounds and phases and their tendency to chemically react to form new
species transform to new physical state and migrate from one spatial location to
another therefore it is the chemical potential differences or gradients that drive
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